TECHNOLOGY FACTSHEET



Retrofit of post-combustion CO2 capture for refineries using MEA solvents Date of factshee 20/05/202 uthor ra Wes Sector 272 Refineries ETS / Non-ETS Type of Technology There are a variety of techniques for post-combustion carbon capture that can be applied to flue gases; this factsheet considers chemical absorption with monoethanolamine (MEA) escription solvents Post-combustion capture does not require any major modifications to the refining process; MEA amine stripping technology is an end-of-pipe technology added to the plant to capture CO2 from existing flue gas streams. The modifications required for CO2 capture are cleaner flue gas (dust filters, NOx removal, additional desulphurisation equipment, etc.); a CO2 capture unit (absorber and stripper columns, heat exchangers, condensers, and a reboiler); and a CO2 compression and dehydration unit. However, the design of the site plays an important role in the cost and feasibility of implementing post-combustion CO2 capture. There are many different flue gas streams on a typical refinery site, and how they are combined in different stacks can vary. Thus the CO2 concentration and flue gas volume per stack will vary in each site as well. As the capture process requires electricity (notably for compression) and steam (mainly for solvent recovery), additional investments are also required to expand the site's utilities. Many refineries have significant excess heat availability, which could reduc the cost and additional energy demand, but the potential to use that heat depends largely on site design and site-specific constraints, so has not been considered in this factsheet. This factsheet is based on literature looking both at hypothetical refinery site configurations and at specific sites. The cleaned flue gas enters the absorber and is brought into contact with the MEA amine solution. About 90% of the CO2 is absorbed into the amine solution (now together referred to as a rich loading solution), and is then pumped to the stripper. In the stripper column, the rich loading solvent is heated with steam from the reboiler (which uses a heat exchanger to transfer heat from external steam to a heat transfer fluid), breaking the chemical bonds between the amine solvent and the CO2, and causing it to release its CO2, creating a relatively pure CO2 stream. The remaining solution (called a lean loading solution), now at a temperature of about 120 degrees C, is pumped back to the absorber to begin the cycle again, first passing through a heat exchanger to preheat the rich loading solution. The CO2 continues to the compressor, which compresses the gas for transport and storage. Pressure in the CO2 pipeline can vary significantly (both liquid and gaseous transport is possible) and has an important impact on total investment cost, operating cost and electricity use. This factsheet onsiders transport at about 110 bar/11 MPa to 140 bar/14 MPa. While the MEA solvent capture technique is can also be applied to flue gases from power plants, there are two major differences when considering a refinery applications. First, the installations require combination of multiple flue gas streams (from various processes and utilities), which leads to higher equipment costs per unit of captured CO2. Second, the final oncentration of CO2 in the flue gases is higher than those of a typical gas-fired power plant (here the refinery concentration is assumed to be around 9% vol. though this is highly ependent on the configuration, where a typical gas-fired power plant may have a concentration in the range of 5-10%vol) Post-combustion carbon capture can be either retrofitted or designed in a greenfield refinery; this factsheet considers a retrofit to an existing refinery. Integrated design could lead to ower costs or higher efficiency. Refinery configurations in the Netherlands vary widely. Key flue gas streams come from utilities (boilers and/or CHPs), steam methane reformers (SMR), fluid catalytic crackers (FCC), distillation processes (atmospheric and/or vacuum). Not all processes are present at all refineries, and thus their shares in the total CO2 emissions of the refinery also vary. Capture fron the SMR unit of a refinery can also be considered separately; capture of CO2 from an SMR only is covered by other factsheets. This factsheet considers several configurations, with capture of CO2 from flue gases from utilities, SMR, FCC, and distillation processes or some subset of those processes. The assumed CO2 concentration in these flue gases is about: utilities ~8%vol, SMR ~24%vol, FCC ~17%vol, atmospheric/vacuum distillation ~11%vol. The average concentration of the flue gas entering the capture unit for these configurations ranges from 6.7% vol to 13.1% vol, with an average of about 9% vol. At lower concentrations, the cost per tonne of captured CO2 rises. The power lant flue gases have slightly lower concentrations than the other sources shown here but typically account for the largest single source of CO2 emissions at a refinery, and therefore are often included in studies of CO2 capture at refineries. (Roussanaly et al., 2017: Ho et al., 2011: Leeson et al., 2014). Furthermore, an average MEA loss of 2 kton/ Mton CO2 is assumed. Based on different cases, losses of MEA may increase due to thermal, oxidative degradation or larger presence of contaminants such as SOx and NOx. These consideration are case specific and left out of scope. TRL leve TRI 8 2020 Post-combustion carbon capture has been demonstrated at full scale in a refinery, and operates commercially using the same capture technique in power plants. Every refinery onfiguration is different, so the specific design and costs will vary, but the basic principle of chemical absorption carbon capture using MEA solvents remains the same. Geological torage of carbon dioxide has also been demonstrated and is commercially available, though there are currently no CCS projects operating in the Netherlands. The Porthos project, which will transport CO2 captured in the port of Rotterdam by pipeline, for storage in retired gas fields in the North Sea, has signed Joint Development Agreements with ExxonMobil and Shell, both of whom operate refineries in the Rotterdam area. Air Liquide and Air Products have also signed Joint Development Agreements. The partners will apply for SDE++ funding for the project. Construction, according to the project timeline, will begin in 2022, and operation will begin in 2024. TECHNICAL DIMENSIONS Value and Range Functional Unit Capacity Mton CO2 captured 1,60 0,50 2,80 Current 2030 2050 Potential % Market share Capacity utlization factor Full-load running hours per yea Aton CO2 Unit of Activity captured/yea Technical lifetime (years) 25,00 Progress ratio Hourly profile Explanation Capacity varies depending on refinery size, configuration, process equipment and utilities. ach of the refineries in the Netherlands has a different configuration and different processes on-site; thus this factsheet is not equally applicable to all Dutch refineries. The total refinin nameplate capacity of the Netherlands is about 67 Mt crude oil intake/year, with 6 refineries ranging from 3.5 Mt/year to 21 Mt/year. More information about Dutch refinery capacity can be found in the MIDDEN report (Oliveira and Schure, 2020). This factsheet is most relevant for a refinery of medium to high complexity with capacity of 220 000 bbl/day throughput or higher, with capture of CO2 from at least three point sources on site. It is not possible to determine the potential or market share of this technology in the future, as it will be highly dependent on policy, subsidies, and CO2 prices, as well as the future of the . Dutch refining sector Itilization factor will likely be similar to the utilization factor of the refinery process equipment. to estimates were available on progress ratio or hourly profile.

COSTS													
Year of Euro		2015											
		Euro per Functional Un	Current			2030			2050				
Investment costs		mln. € / Mton CO2 captured		270,00		300.00	Min – Ma		Max	- Min - Max			
Other costs per year		mln. € / Mton CO2 captured		Min	-	Max	Min	-	Max	Min	-	Max	
Fixed operational costs per year (excl. fuel costs)		mln. € / Mton CO2 captured		10,23	12,09	14,06	Min	-	Max	Min	-	Мах	
Variable costs per year		mln. € / MtCO2 captured		4.05	4,96	4.05	A.47	-	14	6.41-	-		
		The costs of capture cap vary depending on the con		4,90 aplovity and co	-	4,96	IVIIII idorod. cost o	- f conital and atl	iviax	iviin actor and site	-	IVIAX	
Costs explanation		vary significantly based on the layout and configuration of a refinery and processes on site. A key factor is whether it is possible to combine flue gas streams into one stack where capture can be applied. (Leeson et al., 2017). Above, a range of estimates are presented representing different configurations that are relevant for Dutch refineries. Interconnection costs can make up a large share of the total investment costs (about 20-30% according to Roussanaly et al., 2017). Note that utility investment costs can also be an important share of the total investment costs (in the case of Roussanaly et al., 2017; around 20% for an additional natural gas CHP on site); costs for utilities are excluded from the investment costs glove above. Variable costs include chemicals and catalysts, process water, and waste disposal. (Roussanaly et al. 2017). Cost estimates are often presented on an annualized basis, per tonne of CO2 captured or avoided, including fixed and variable O&M costs. These costs are not directly comparable to the investment costs presented above, which are overnight capital investments per unit of capacity. Note that cost per tonne of CO2 captured differs from the cost per tonne of CO2 avoided (which takes into account the energy penalty - CO2 emitted in generating the additional energy needed for capture and compression of CO2 - and is therefore higher). For example, for medium-to-high complexity refineries, estimates range from \$90 to 120 per tonne of CO2 captured (about 83-116 EUR 2015) (van Straelen et al., 2010) to 5160 to 180 per tonne of CO2 avoided (about 133-156 EUR 2015) (Roussanaly et al., 2017; Gale, 2017). Another study of post-combustion CO2 capture in oil refineries in Australia finds a cost of about \$87/t CO2 captured (about 88 EUR 2015) (Han et al., 2011); however, this study considers captured only from the combined flue gas of the process heaters. This flue gas has a lower overall concentration of CO2 (~8%) than what is considered in the other studies, and alos requires different on-f											
ENERGY IN-	AND OUTPUTS												
		Energy carrier	Unit		Current			2030			2050		
		Main output:			3,68			-			-		
		Steam	PJ	3,30	-	4,40	Min	-	Max	Min	-	Max	
Energy carrie	ers (per unit of main	Electricity	PJ	0.45	0,61	0.62	Min	-	Max	Min	-	Max	
output) Energy in- and Outputs explanation			PJ	0,+5	-	0,02	1.4i-	-	100A		-	h d mu	
			PJ	IVIIN	-	IVIAX	IVIII	-	IVIAX	IVIIN	-	IVIAX	
		Steam demand for CO2 capture can be met by on-site utilities or can be purchased from an off-site steam generator. CAPK2 considered above, from Roussnalv et al. (2017) and Gale (2017), excludes the cost of expansion of utilities to meet the additional steam demand for CO2 capture. Some sites may already have sufficient steam generation capacity on-site to meet the extra demand; in this case, additional fuel will be consumed. Steam demand is shown here, rather than fuel, to provide a generic case relevant to most refineries. Note that most of the electricity demand is related to compression of CO2. This depends highly on the required pressure for CO2 transport, and thus is highly site- and project-dependent. A range is presented here, with low range estimates representing lower transport pressures (~20bar) and higher electricity demand representing higher pressures (~110bar).											
MATERIAL F	LOWS (OPTIONAL)						1						
Material flows		Material	Unit		Current			2030			2050		
		MEA solvent (make-up)	kt		2,09			-			-	-	
				2,09	-	2,09	Min	-	Max	Min	-	Max	
				Min	-	Max	Min	-	Max	Min	-	Max	
Material flow	ws explanation												
EMISSIONS	(Non-fuel/energy-rel	ated emissions or emissions reduction	ons (e.g. CCS)	-									
Emissions		Substance	Unit		Current			2030			2050		
		CO2 captured	Mton CO2-eq	-1.00	-1,00	-1.00	Min	-	Max	Min	-	Мах	
					-			-			-		
				Min	-	Max	Min	-	Max	Min	-	Max	
				Min	-	Max	Min	-	Max	Min	-	Max	
				Min	-	Max	Min	-	Max	Min	-	Max	
Intere is a wide range of emitted CO2 in these refineries. The variety of configurations considered have different total baseline CO2 emissions, as well as different shares of avoic captured emissions, based on the different point sources of CO2 where capture techniques are applied. For the remaining emissions, have and use in the different point sources of CO2 where capture techniques are applied. For the remaining emissions, the main value (0.38t CO2 emitted) represents a configurations considered have different total baseline CO2 emissions, as well as different shares of avoic captured emissions, based on the different point sources of CO2 where capture techniques are applied. For the remaining emissions, the main value (0.38t CO2 emitted) represents a configuration where capture is applied on utilities, FCC, SMR, and distillation units at a medium- or high-complexity refinery. The high end of the range (1.82 tCO2 emitted) represents a configure sources on site which are not suitable for capture; this means that the overall capture rate for a re (CO2 (85-50%), is not achievable for the whole refinery, as there are small CO2 sources on site which are not suitable for capture; this means that the overall capture rate for a re lower, depending on the particular configuration. (Roussanaly et al., 2017) (Ornaheim et al 2015). Overall capture rates for the data presented in this factsheet range from abou about 75%, including CO2 from on-site utilities.									voided and resents cases onfiguration gle stream of a refinery is bout 35% to				
OTHER													
Parameter		Unit		Current			2030			2050			
				Min	-	Max	Min	-	Max	Min	-	Мах	
				Min	-	May	Min	-	Max	Min	-	May	
				Min	-	May	Min	-	Max	h dia	-	May	
					-	IVIUX	iviini 	-	INIUX	IVIIII	-	IVIUX	
Explanation				IVIIN	-	IVIAX	Min	-	IVIAX	IVIIN	-	Max	
Explanation		L											
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